

October 25, 2023

# Vapour Pressure of Xe Isotopes & The XeStill Project

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# Motivation

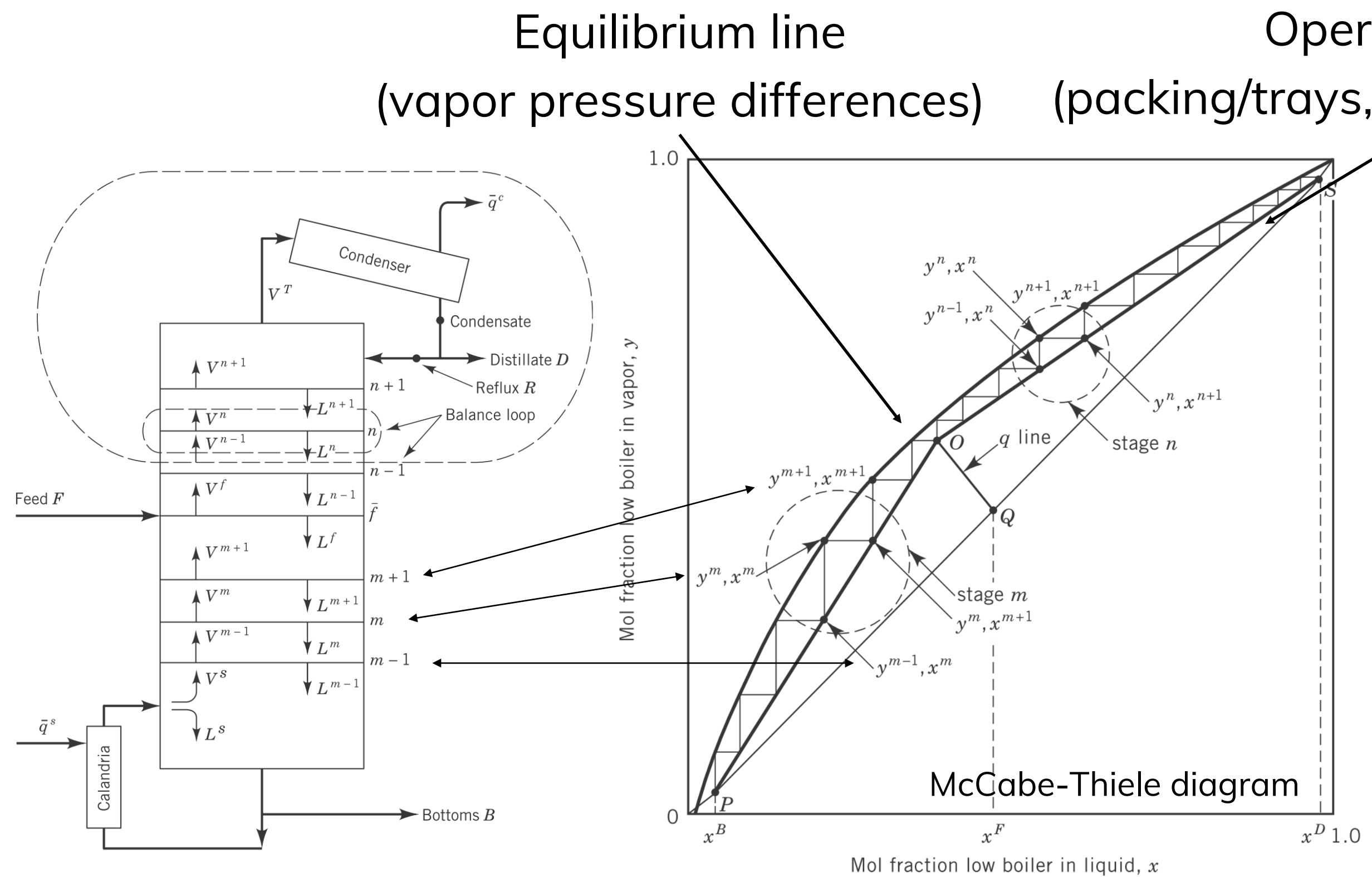
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- In the past, understand the triple point (TP) of xenon to be used as a **standard temperature**

*Tew, W. L., Intl J of Thermophysics (2008), 29*

- Fail to produce credible data
- More recently, need of **xenon enrichment** for  $0\nu\beta\beta$  searches
  - Natural xenon: ~80%  $^{128-132}\text{Xe}$ , 8.9%  $^{136}\text{Xe}$
  - Only approach employed to enrich is via centrifugation: costly power consumption
  - Distillation uses gravity and isotopic differences in vapour pressures
- Cryogenic distillation was used to produce heavy water in the SNO experiment

# Concepts of (Cryo) Distillation



Equilibrium line  
(vapor pressure differences)

Operation line  
(packing/trays, number of stages)

(Isotopic) separation parameters:

- Number of stages: N
- Height equivalent theoretical plate:
  - HETP = column height / N
- Vapor pressure differences
- Vapour pressure isotopic effect:

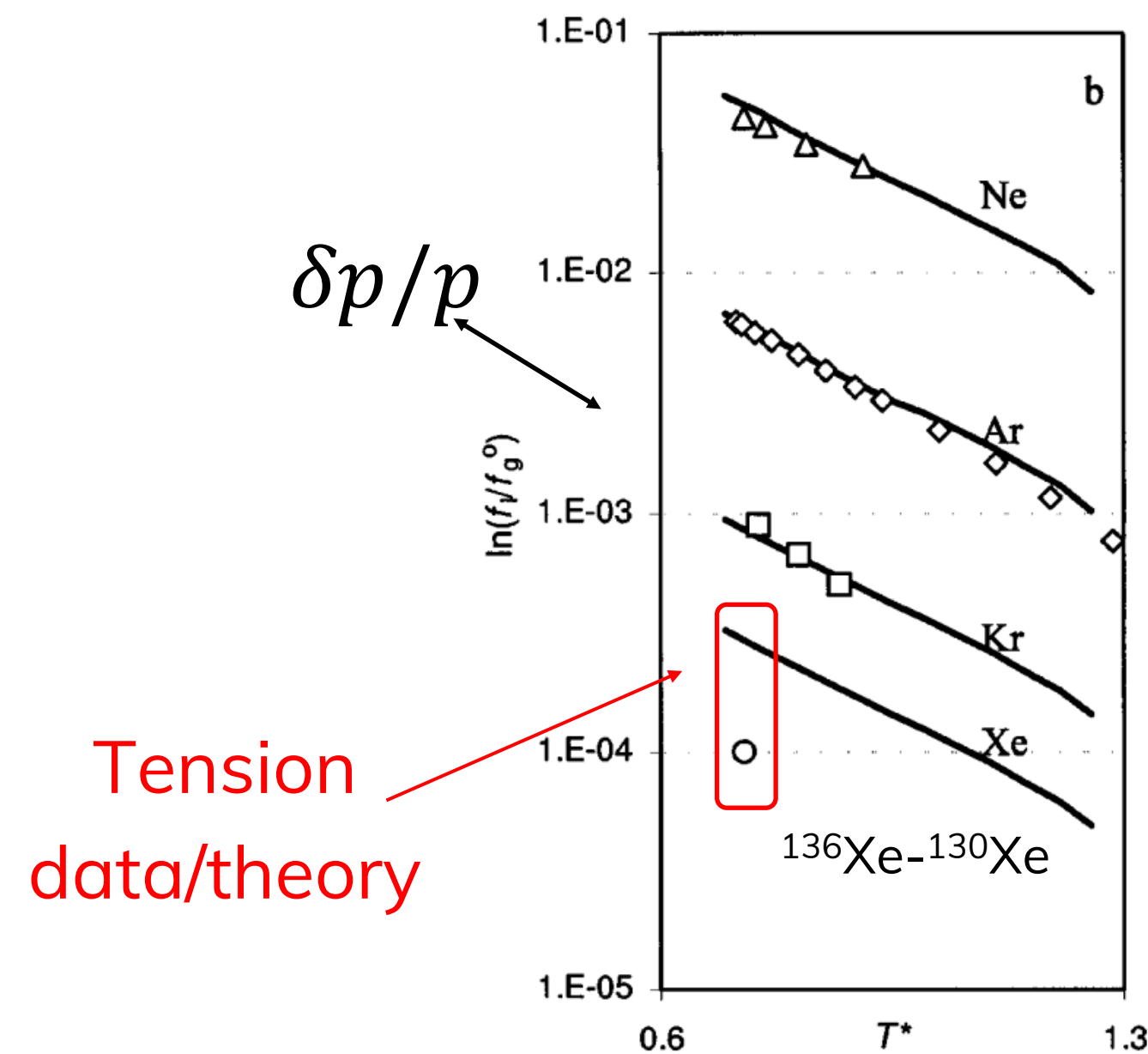
$$\frac{\delta p}{p} = (p_h - p_l) / p$$

$$\text{VPIE} := \ln\left(\frac{p_l}{p_h}\right) \approx \ln\left(\frac{f_c}{f_s}\right) \approx \ln(\alpha_{l-h})$$

*J. Fair and H. Kister, "Distillation",  
Kirk-Othmer Encyclopedia of Chemical Technology (2018)*

# Existing Data / Models

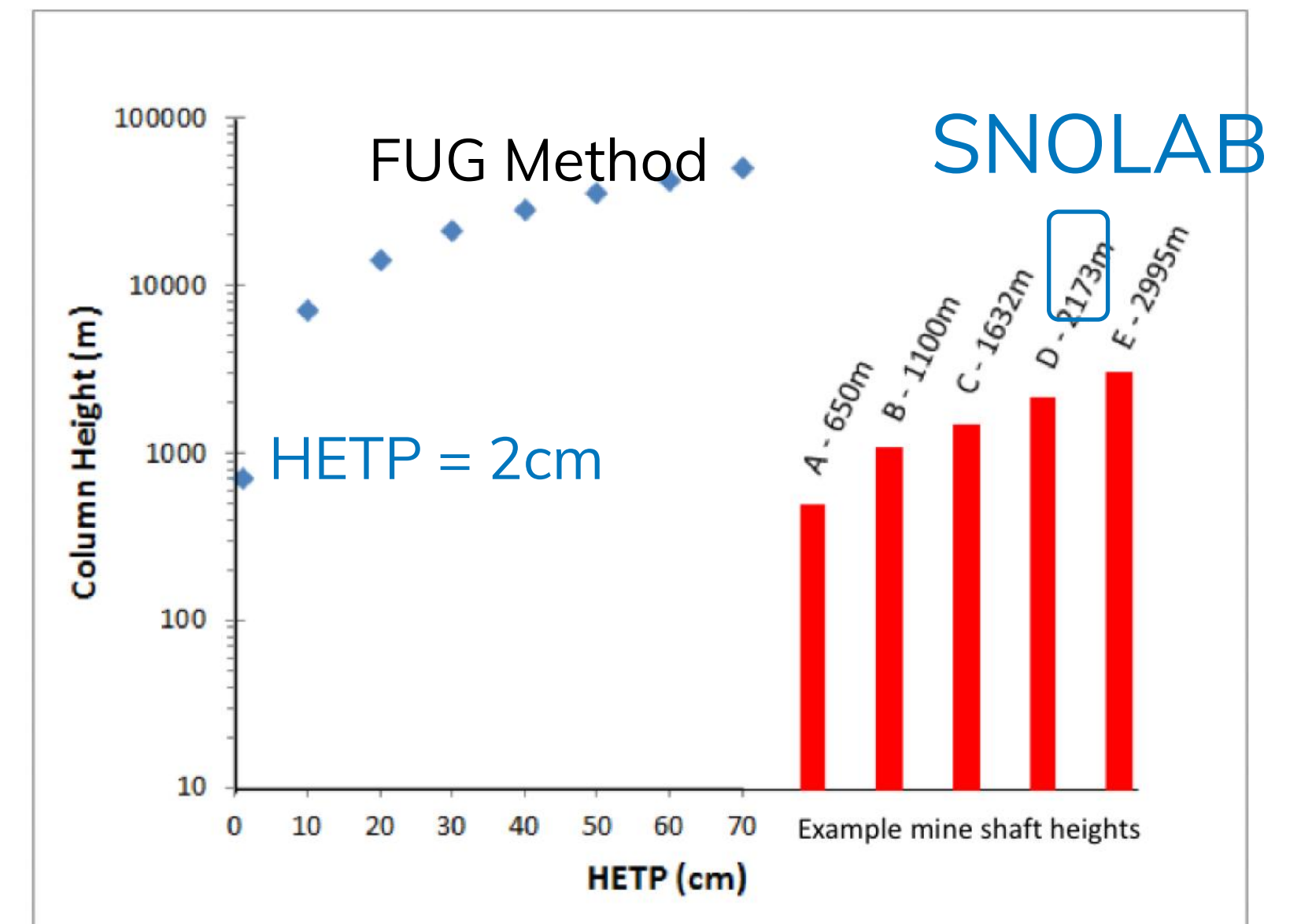
- Data exists for some fluid isotopes at various temperatures:  $^{22}\text{Ne}$ - $^{20}\text{Ne}$ ,  $^{40}\text{Ar}$ - $^{36}\text{Ar}$ ,  $^{84}\text{Kr}$ - $^{80}\text{Kr}$
- Two thermodynamic models describe these fairly well



$$\ln \alpha_{l-h} = \frac{\hbar^2}{24 \varepsilon \cdot \sigma^2} \frac{\Delta m}{m_l m_h} \left( \frac{A}{T^{*2}} + \frac{B}{T^*} + C \right) \propto \Delta m / m^3$$

Theoretical prediction is 1/3 of Kr:

$\delta p/p \sim 0.1\% \rightarrow$  sensitivity need of 0.01%



H.O. Back et al 2017 JINST 12 P09033

J. Canongia-Lopes et al J. Chem. Phys. 118, 5028 (2003)

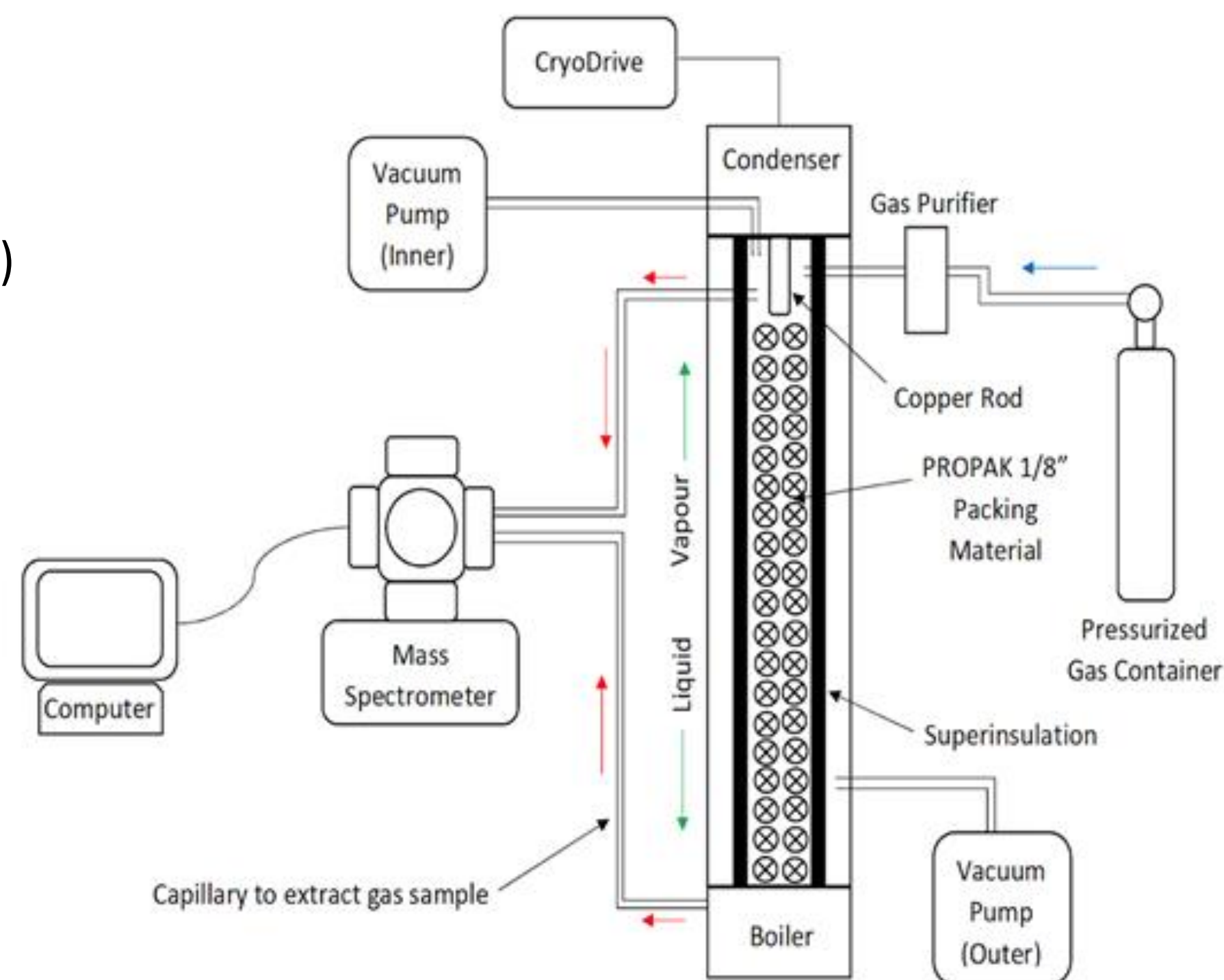


# Still Setup @ Carleton

- 1.8 m high cryogenic still
- 25 mm inner diameter vacuum insulated reflux column filled with PROPAK (random) packing

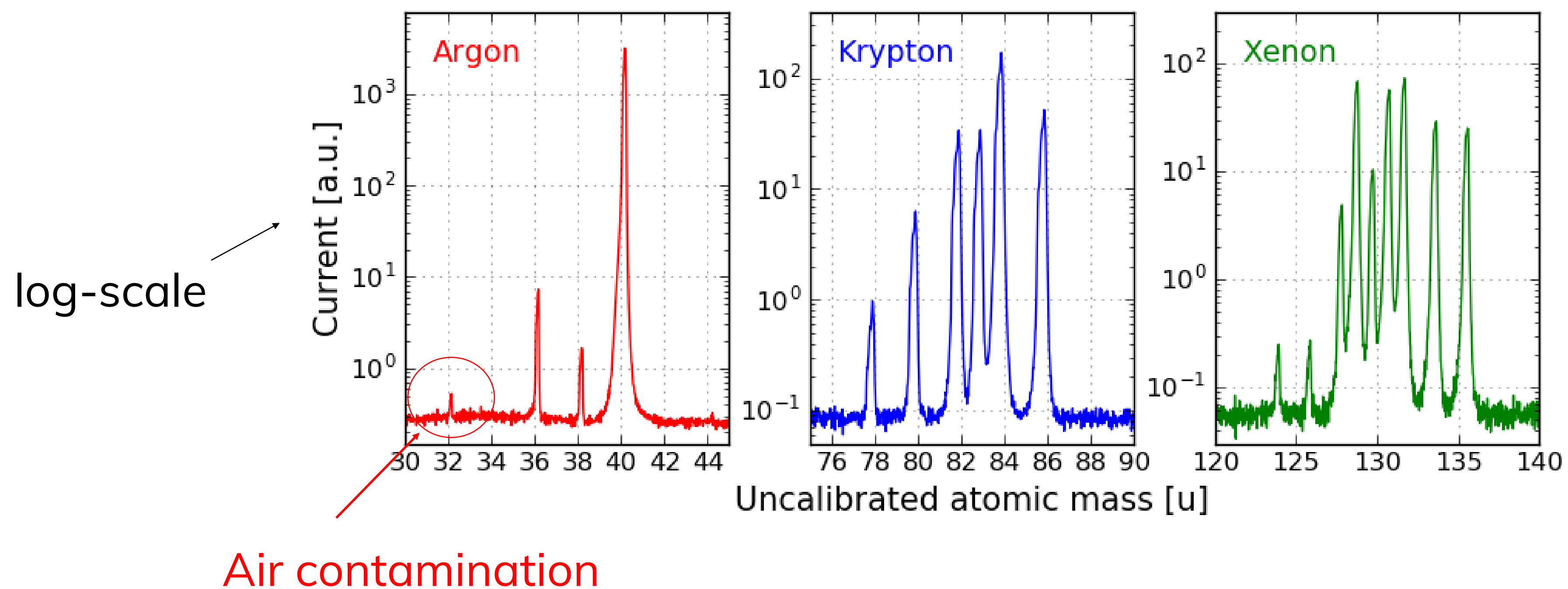


- One condenser placed at the top and re-boiler at the bottom
- 100% reflux
- Pressure of 105 kPa



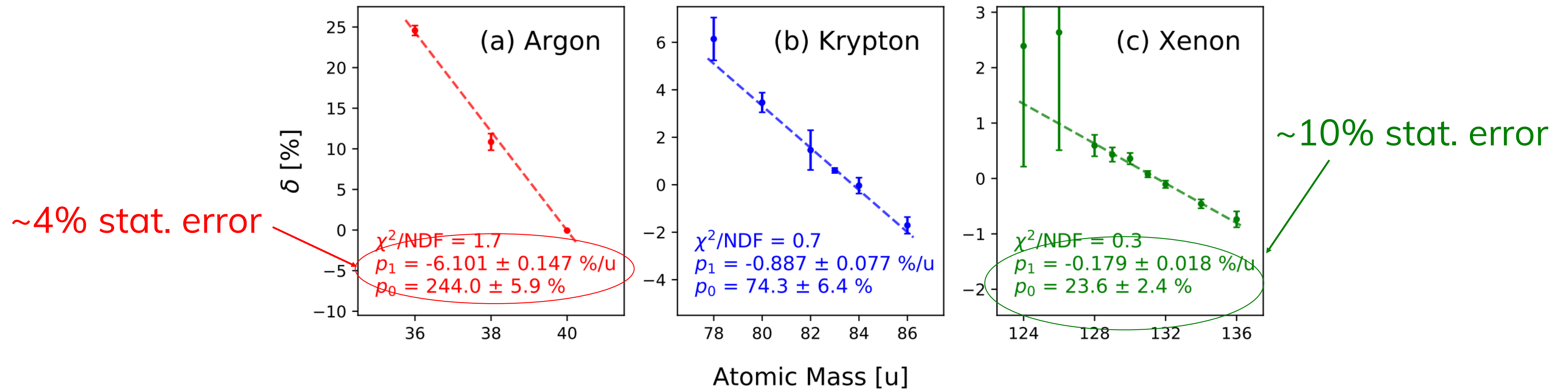
# Mass Spectra

- EXTREL quadrupole: 19 mm bore, mass resolution of 1:2000
- Focus on single ions





# Analysis Procedure



$$\delta = \frac{f_{i,top} - f_{i,bot}}{(f_{i,top} + f_{i,bot})/2} \times 100$$

Measurement of relative enrichment  
top-bottom per isotope

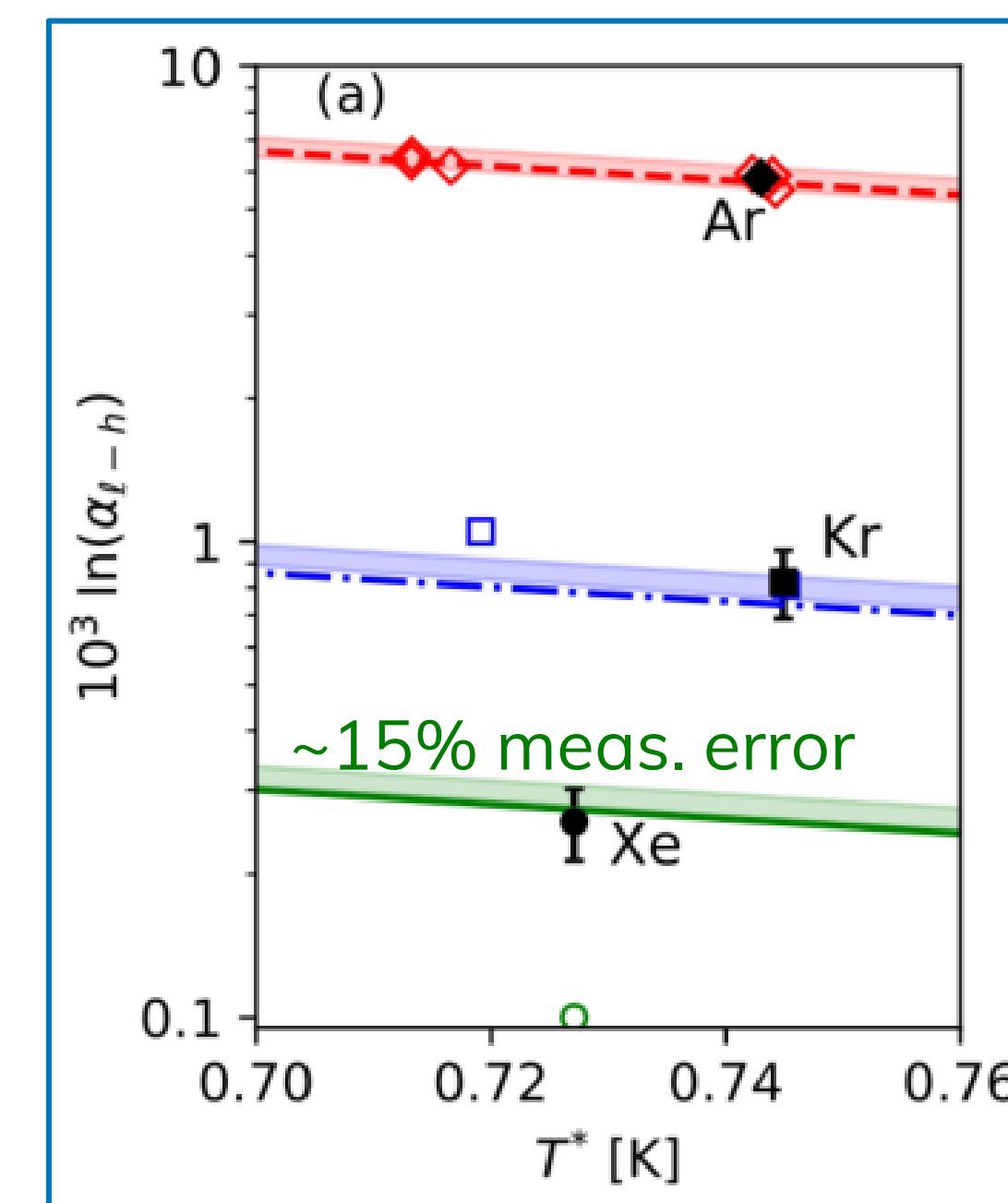
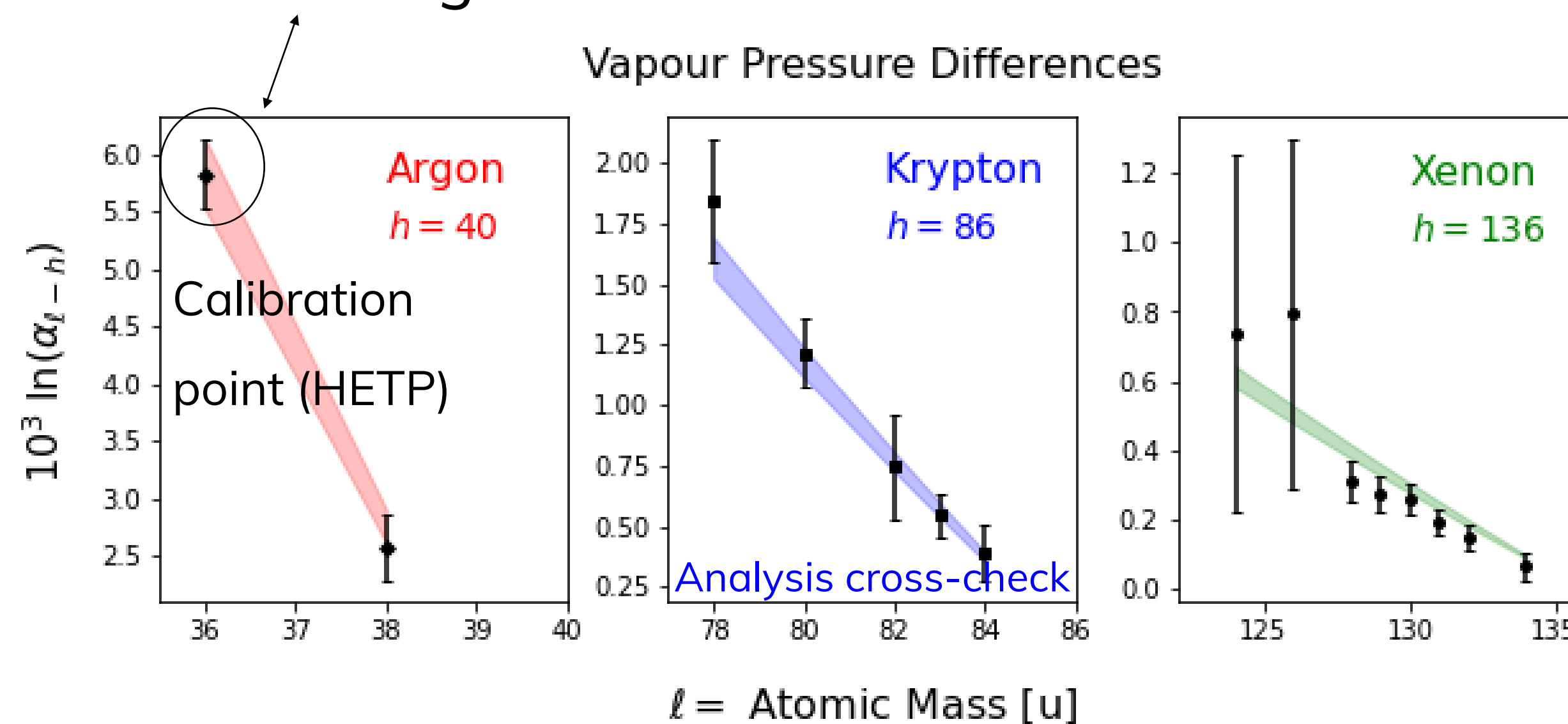
# Initial Results

100% reflux, Fenske equation:

$$N(t) = \frac{\ln((\delta_{l-h})(t))}{\ln((\alpha_{l-h})(t))} \quad \delta_{l-h} = \frac{(1 + \delta_l/200)/(1 - \delta_l/200)}{(1 + \delta_h/200)/(1 - \delta_h/200)}$$

HETP =  $4.23 \pm 0.25$  cm

N =  $42.5 \pm 2.5$  stages



A. Alamre et al, ACS Omega 2020, 5, 45, 28977–28983



# Still @ SNOLAB

Located in the Cryopit, 8x increase, ~14m high

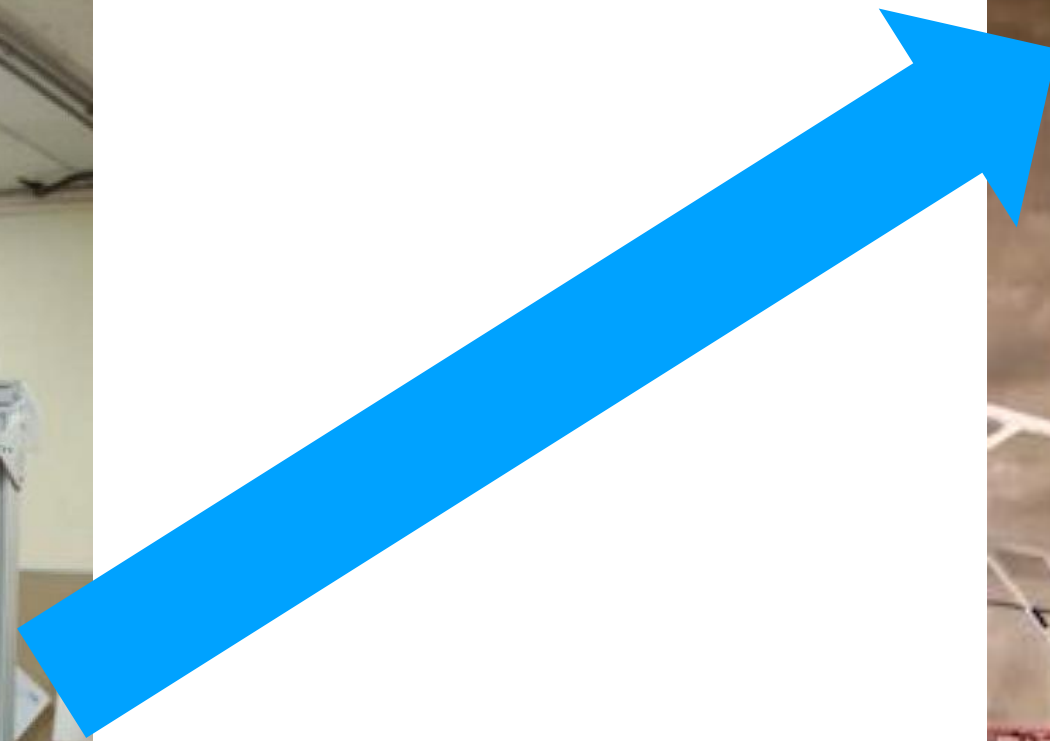
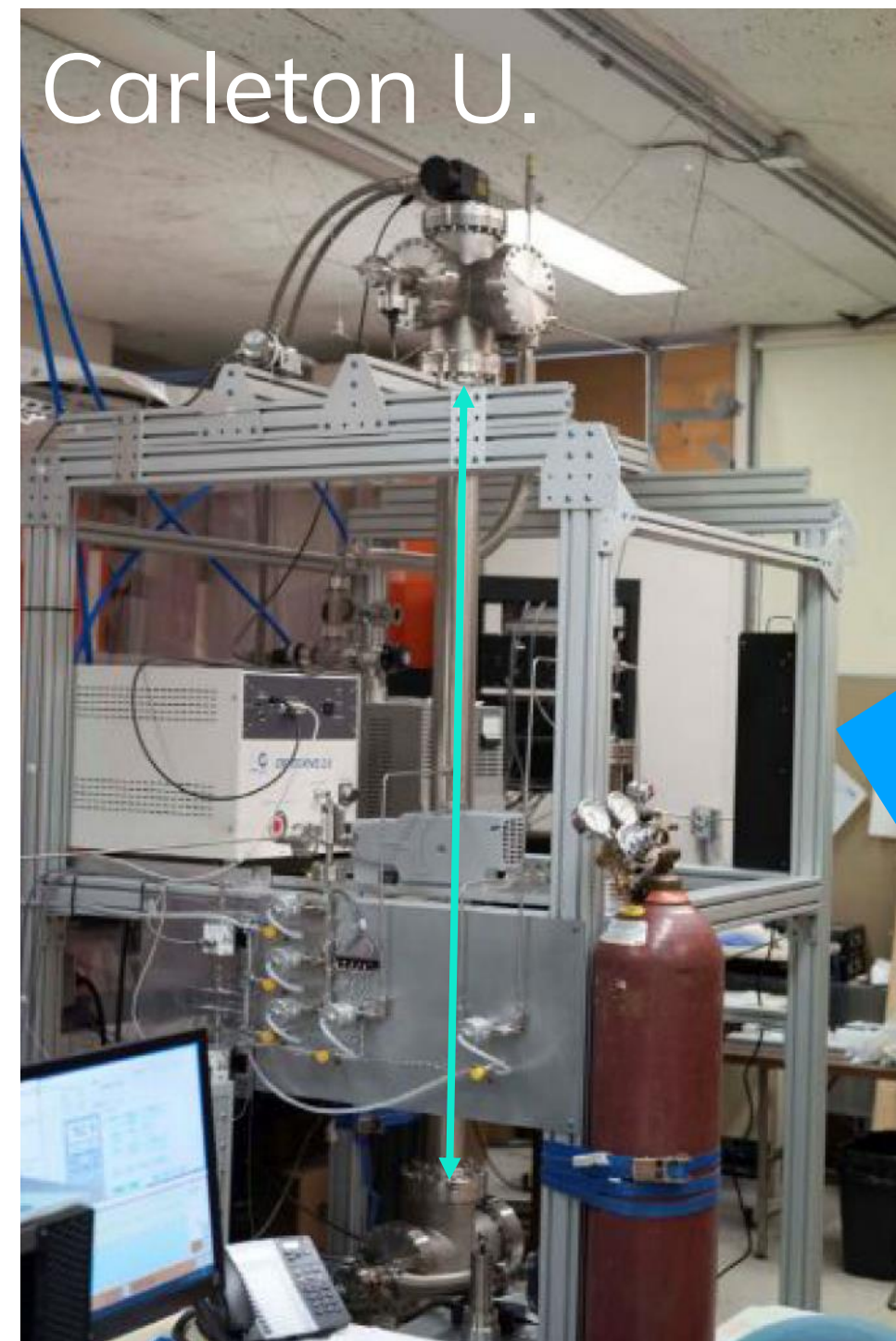


Photo courtesy of Andrea Pocar

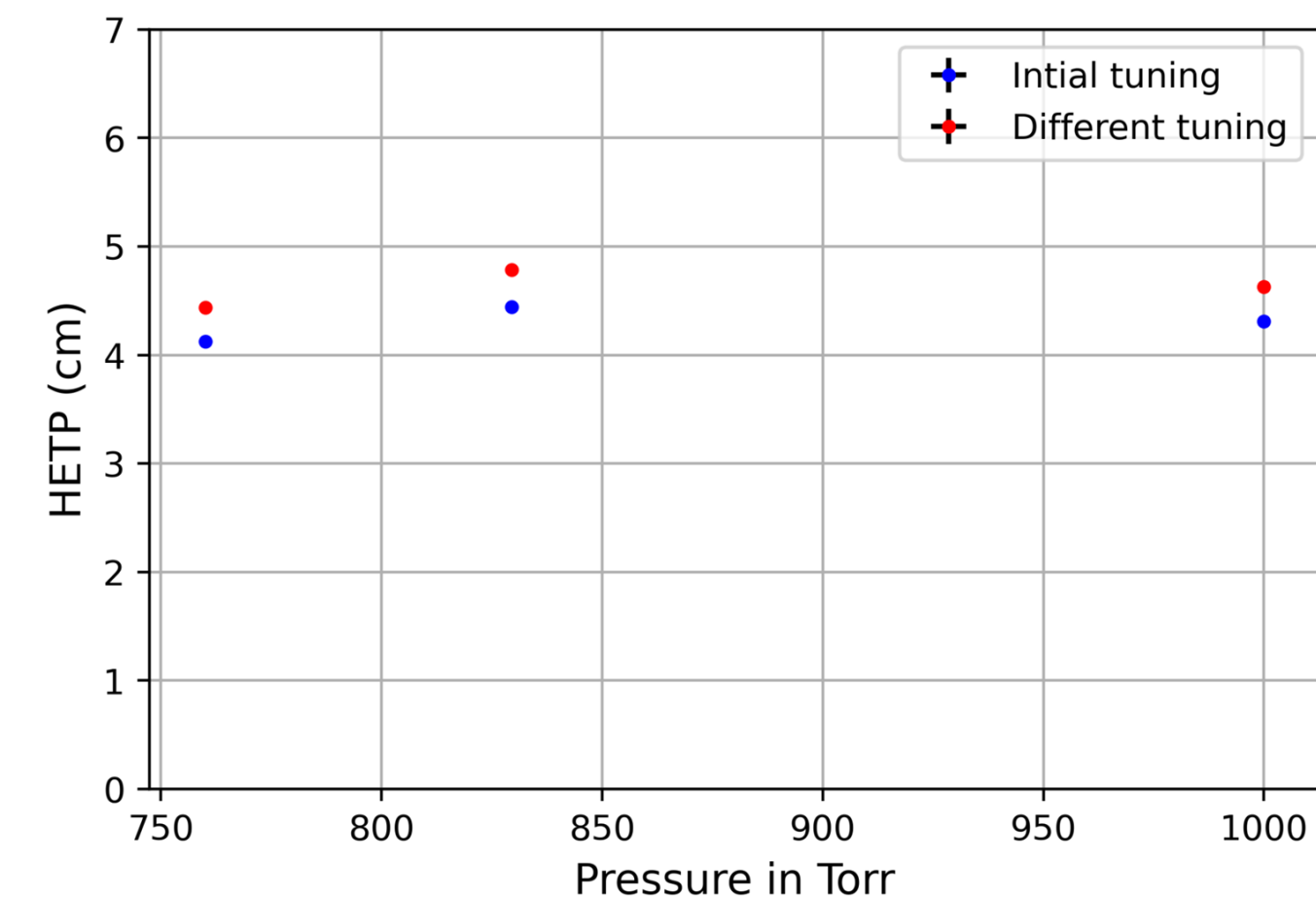
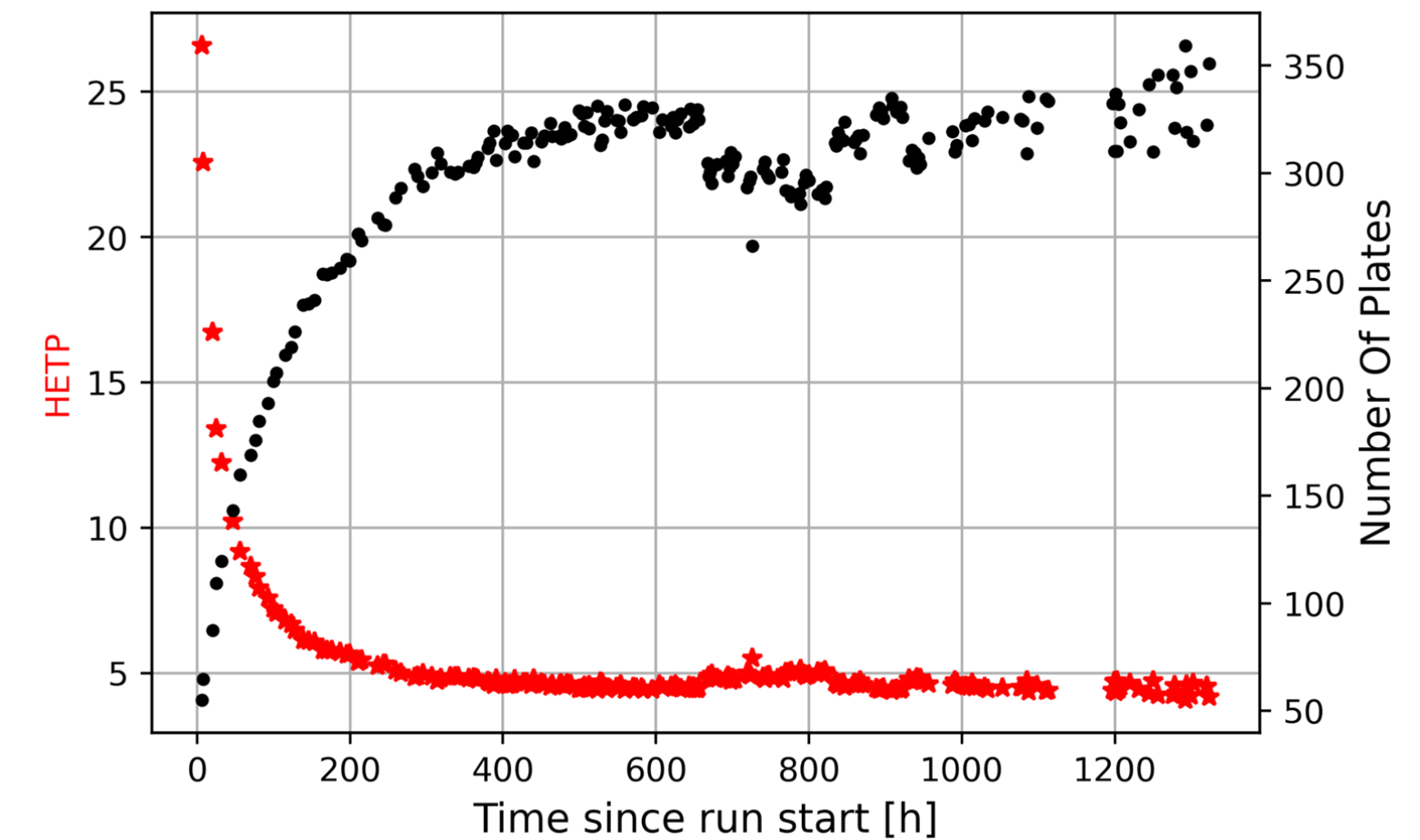


# SNOLAB Results (so far)

- MSc thesis (A. Emara, 2023) measured HETP for argon

SNOLAB vs Carleton  
 Number of stages:  $328.7 \pm 0.9$  (vs  $42.5 \pm 2.5$ )  
 HETP:  $4.29 \pm 0.13$  cm (vs  $4.23 \pm 0.25$  cm)

- Data for krypton obtained this summer 2023, on-going analysis, validation of HETP
- Running for 2 years, weeks or months at a time
- Challenges running underground:
  - Long evolution time (~month vs week)
  - Operations, hardware maintenance
  - Precision measurements, systematics





# XeStill Team

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Carleton University:

- David Sinclair (PI)

University of Windsor

- Caio Licciardi (PI), Drake Wickman (MSc student)

SNOLAB:

- Steffon Luoma (scientist), Ashley Mathewson (project engineer)

*Former members:*

- *Jason Anstey, Jeff Mason (technical staff), Vance Strickland (project engineer)*
- *Abo-Bakr Emara (Carleton, MSc graduate), Amal Alamre (Carleton, MSc graduate), Brandon Death (Carleton, undergrad thesis)*
- *Dr. Ibtisam Badhrees*
- *SNOLAB scientists: Erica Caden, Jeter Hall, Steve Linden, Brian Morissette*





# Conclusions

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Cryogenic distillation is a potential approach for xenon enrichment

Distillation relies on vapour pressure differences between xenon isotopes

- Very small, requires precision measurements for optimal design

- Estimated column heights ~100-1000s meters

Delivered results at 15% error, consistent with existing expectations

Working on precision measurements with higher column at SNOLAB's Cryopit

- Successful calibration measurement

- Validation is on-going

- Xenon measurements expected in Winter 2024